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Thickness-dependent properties of ultrathin bismuth and antimony chalcogenide films formed by physical vapor deposition and their application in thermoelectric generators



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ABSTRACT

In this work, a simple cost-effective physical vapor deposition method for obtaining high-quality Bi₂Se₃ and Sb₂Te₃ ultrathin films with thicknesses down to 5 nm on mica, fused quartz, and monolayer graphene substrates is reported. Physical vapor deposition of continuous Sb₂Te₃ ultrathin films with thicknesses 10 nm and below is demonstrated for the first time. Studies of thermoelectrical properties of synthesized Bi₂Se₃ ultrathin films deposited on mica indicated opening of a hybridization gap in Bi₂Se₃ ultrathin films with thicknesses below 6 nm. Both Bi₂Se₃ and Sb₂Te₃ ultrathin films showed the Seebeck coefficient and thermoelectrical power factors comparable with the parameters obtained for the high-quality thin films grown by the molecular beam epitaxy method. Performance of the best Bi₂Se₃ and Sb₂Te₃ ultrathin films is tested in the two-leg prototype of a thermoelectric generator.

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1. Introduction and background

Thermoelectric materials and based technologies, enabling the direct conversion of heat to electricity, are expected to play an important role in solving one of the biggest global challenges, which is increase of energy efficiency of domestic and industrial processes by waste heat capturing and conversion to useful energy. A simplest thermoelectric generator (TEG) for heat-to-electricity conversion consists of two semiconductors with different (n- and p-) types of electrical conductivity, joined at their ends. The performance of thermoelectric materials and devices is characterized by a dimensionless figure of merit ZT, defined as $S^2\sigma T/k$, where S is the Seebeck coefficient of the material, σ is its electrical conductivity, T is absolute temperature, and k is thermal conductivity of the material. $S^2\sigma$ is usually referred as the power factor (PF) of

* Corresponding author. E-mail address: donats.erts@lu.lv (D. Erts). thermoelectric material. The common approaches in enhancement of ZT is maximizing of PF and/or reduction of thermal conductivity of the material [1]. However, interdependency of electrical and thermal conductivities hampers breakthrough enhancement of ZT of materials.

Bismuth and antimony chalcogenides (Bi₂Se₃, Bi₂Te₃, Sb₂Te₃), layered narrow—band-gap semiconductors with recently discovered 3D topological insulator (TI) properties [2–4], are widely known as the best near—room temperature thermoelectric materials with n-type (Bi₂Se₃, Bi₂Te₃) and p-type conductivity, useful for domestic waste heat conversion applications. These materials consist of charge neutralized layers (QLs) bonded together by van der Waals interaction. Each QL is formed by five covalently bonded atomic sheets (for example, Se–Bi–Se–Bi–Se) and is ~1 nm thick.

Recently, it was predicted that quantum confinement introduced by downsizing of the thermoelectric materials may result in enhancement of ZT of thermoelectric materials [5,6]. For example, it has been shown theoretically that the multiple sub-band structure, small band gap, and vicinity of materials with TI properties to

a metallic state associated with an electronic topological transition are responsible for their superior TE performance [7]. Theoretical calculations performed for TI bismuth chalcogenide films [8,9] proved that the reduction of the thickness of thin film below 10 nm will result in noticeable quantum tunneling between the top and bottom surfaces and inducing of hybridization band gap near the Dirac point, resulting in improvement of existing ZT of the material by nearly an order of magnitude due to the disengagement of electrical and thermal conductivities of this material. At the same time, the other research groups predicted that the parallel contributing of coexisting in such TI systems surface and bulk states to the thermoelectric properties of the TI material may tend to cancel each other out, and ZT cannot be enhanced dramatically; however, some improvement of thermoelectrical properties of material could still be observed at optimal positions of Fermi levels [10]. Experimental investigation of properties of nanostructured materials of the bismuth and antimony chalcogenide family also showed different results. For example, properties of nanostructured n-type BiTe showed coexistence of distinct topological quantum phases in this material resulting on potential decoupling of thermal and electronic transport properties, which is the fundamentals for efficient thermoelectrics [11].

On the other hand, while the theoretically proposed thicknesses for the hybridization gap opening, leading to the disengagement of electrical and thermal conductivities and consequent improvement of ZT of the material are below 6 nm and 3 nm for Bi₂Se₃ and Bi₂Te₃, respectively [10], the experimental investigation of grown by molecular beam epitaxy (MBE) ~5 nm thin (Bi_{1-x}Sb_x)₂Te₃ and ~5-30 nm thin Bi₂Se₃ films did not show any significant improvement from the bulk values of ZT [12,13]. Regarding Sb₂Te₃ thin films, while theoretical studies predicted significant sensitivity of Seebeck coefficient and electrical conductivity of these thin films to the thickness reduction down to 2 nm, the experimental measurements were performed only for the Sb₂Te₃ thin films with thicknesses down to 30 nm fabricated by atomic layer deposition and revealed increase of the electrical conductivity but at the same time decrease of the Seebeck coefficient, resulting to reduction of thermoelectric efficiency of the material related to the contribution of surface states [14]. The method for gaining a thermoelectrical benefit from the Sb₂Te₃ thin film proposed by the aforementioned research was gapping the surface states and tuning the charge carrier concentration. However, the theoretically proposed thickness for hybridization gap opening in Sb₂Te₃ is 4 nm [10], and, to the best of our knowledge, the experimental research of the thermoelectric properties of Sb₂Te₃ thin films with thicknesses below 30 nm has not been yet performed.

This work is focused on development of a simple and costeffective physical vapor deposition (PVD) method of synthesis of Bi₂Se₃ and Sb₂Te₃ ultrathin films of thicknesses down to 5 nm on different (natural mica, quartz, graphene) substrates. Mica is recognized as an excellent non-electrically conductive substrate for the growth of layered materials as bismuth and antimony chalcogenides. Mica is a chemically inert substrate with atomically flat surface ideal for ultrathin films growth by van der Waals epitaxy. For example, the minimal reported thicknesses of the separate Bi₂Se₃ nanoplates PVD-deposited on synthetic fluorophlogopite mica were as small as 2–3 nm [15]. In this work, natural muscovite mica is used, which is much cheaper alternative to the synthetic mica. However, the drawbacks of mica substrates for thermoelectric measurements are its relatively high thermal conductivity, masking the real thermal conductivity of ultrathin films deposited on its surface. A good alternative to mica substrates for the thermoelectric measurements is fused quartz substrates, having low thermal conductivity. However, deposition of the ultrathin films on quartz substrates is challenging because of the presence of variety

of chemical and mechanical defects on its surface. Finally, graphene is a very attractive substrate for the epitaxial growth of ultrathin Bi₂Se₃ and Sb₂Te₃ films because of the small lattice mismatch with these materials (2.9% [16] and 4% [17] for Bi₂Se₃ and Sb₂Te₃, respectively). Moreover, because of high electrical conductance graphene has a potential to be used simultaneously as the substrate for ultrathin films deposition and as the electrode for the measurements. Electrical and thermoelectric properties of Bi₂Se₃ and Sb₂Te₃ ultrathin films of thicknesses ranging from 5 to 15 nm are investigated, discussed, and compared with the properties of thin films grown by the MBE method. Performance of Bi₂Se₃ and Sb₂Te₃ ultrathin films with the best thermoelectric performance is tested in a prototype of the two-leg TEG.

2. Results and discussion

2.1. Synthesis

Previously, our group developed the PVD method for the Volmer–Weber growth of Bi₂Se₃ thin films on mica, glass, and graphene surfaces, using Bi₂Se₃ compound as a source material [18–20]. However, the minimal thicknesses of thin films grown by this method exceeded 50 nm. In this work, to achieve deposition of thinner thin films, the Bi₂Se₃ source material was replaced with separate Bi and Se sources. The use of separate sources allowed to differentiate their evaporation rates by placing them in different temperature zones as shown in Fig. 1a.

The substrate was placed in the temperature region ensuring formation of stoichiometric Bi₂Se₃ [18,20]. Presumably, the layerby-layer growth of Bi₂Se₃ ultrathin films when separate Bi and Se sources are used instead of Bi₂Se₃ powder can be explained by specifics of evaporation and deposition processes of different source materials. The evaporation of Bi₂Se₃ at temperatures above 427 °C has dissociative nature: $Bi_2Se_3 \leftrightarrow 2BiSe(gas) + 1/2Se_2(gas)$ [21]. Because of the higher heat of formation of BiSe (\sim 167 kJ mol⁻¹) than Bi_2Se_3 (~140 kJ mol⁻¹) [22], the adsorbed on the substrate BiSe molecules diffuse across the surface until they reach the energetically favorable position. This process results in nucleation and further growth of Bi₂Se₃ islands [18]. In case of the use of separate Bi and Se sources, heat of vaporization of Se is nearly five times smaller than of Bi (37.7 and 179 kJ mol⁻¹, respectively) [22]. The deposition process starts with evaporation and impinging of Se atoms on the whole substrate surface followed by evaporation of Bi and formation of the Bi₂Se₃ compound over all the surface of the substrate. Using this method, Bi₂Se₃ ultrathin films with thicknesses 5, 6, and 15 nm on mica substrates were prepared.

For the deposition of Sb₂Te₃ ultrathin films, the source material was placed in the center of the furnace hot zone (Fig. 1b). The mechanism of the Sb₂Te₃ growth differs from that of Bi₂Se₃, where Bi is mainly in the atomistic form. The dissociative evaporation of the Sb₂Te₃ compound can be described by the equation Sb₂Te₃ $\leftrightarrow 1/$ $2Sb_4(gas) + 3/4Te_2(gas)$ [21]. Because of the high stability of Sb_4 tetramers and their lower sticking coefficient to the substrate than Sb₂Te₃ [21,23], the Sb₂Te₃ deposition process occurs in three steps. Initially, the optimal for Sb₂Te₃ nucleation substrate temperature must be lower than the threshold for Sb₂Te₃ nucleation (~190 °C), but high enough to avoid Sb clustering. In the developed synthesis method, this condition is fulfilled when the temperature in the center of the quartz tube reaches 405-410 °C and the nucleation of Sb₂Te₃ on the substrate takes place and may be visually observed. However, after the nucleation stage the substrate temperature can be increased by 10–20 °C to improve the Sb₂Te₃ film quality [21,23]. At this step, the further increase of the temperature in the furnace tube center up to 415 or 420 °C and consequent increase of the substrate temperature resulted in formation of 5 nm and 7 nm thin

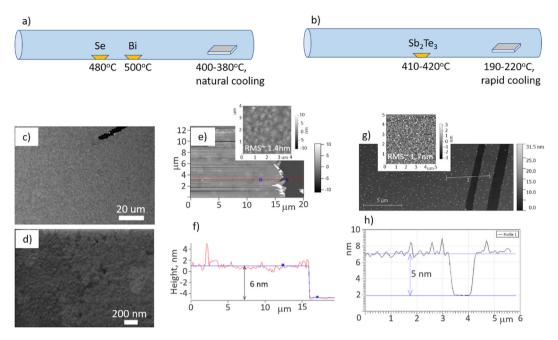


Fig. 1. a—b) Schematics of the experimental setup for Bi₂Se₃ (a) and Sb₂Te₃ (b) ultrathin films synthesis; c—f) The SEM (c,d), AFM (e) image and thickness profile (f) of 6 nm thin Bi₂Se₃ film; g—h) The AFM (g) image and thickness profile (h) of 5 nm thin Sb₂Te₃ film. AFM, atomic force microscope; SEM, scanning electron microscope; RMS, root mean square roughness.

Sb₂Te₃ films, respectively. This step is followed by immediate rapid cooling of the substrate in the inert atmosphere to preserve the Sb₂Te₃ ultrathin film of desired thickness.

The described methods of deposition of Bi_2Se_3 and Sb_2Te_3 films are effective for obtaining of ultrathin films with thicknesses down to 5 nm on the mica surface (Fig. 1 e-h) and down to 8–10 nm on fused quartz and graphene substrates (Fig. S1).

2.2. Structural characterization

Low- and high-magnification scanning electron microscope (SEM) images (Fig. 1c,d; an artificial scratch in Fig. 1c is introduced for the thickness measurements) show that the obtained ultrathin films are uniform and fully fused. Surface scans (Fig. 1 e,g) and thickness profiles (Fig. 1 f,h) performed by the atomic force microscope (AFM) for 6 nm thin Bi₂Se₃ film and 5 nm thin Sb₂Te₃ film show that these ultrathin films have smooth surface with the root mean square roughness (RMS) ~1.4 nm and ~1.7 nm, respectively (insets in Fig. 1 e,g). Energy-dispersive X-ray spectroscopy (EDX) measurements of the ultrathin Bi₂Se₃ and Sb₂Te₃ films deposited on quartz substrates confirmed within the standard deviations the stoichiometric composition (40 [Bi, Sb]: 60 [Bi, Te] at %) of these films. These results were supported by the X-ray photoelectron spectroscopy (XPS) analysis and Raman spectroscopy of ultrathin Bi₂Se₃ and Sb₂Te₃ films.

An example of XPS spectra of the 6 nm thin $\rm Bi_2Se_3$ film is shown in Fig. 2 a,b. The XPS analysis showed the presence of Bi and Se in the material. In high-resolution spectra, the peaks corresponding to $\rm Bi4f_{7/2}$ and $\rm Bi4f_{5/2}$ are located at 158.5 eV and 163.8 eV, while the positions of $\rm Se3d_{5/2}$ and $\rm Se3d_{3/2}$ peaks are located at 53.8 eV and 54.6 eV, respectively.

The analysis of the presented in Fig. 2 c,d representative XPS spectra of the 5 nm thin Sb_2Te_3 film showed the presence of Sb and Te in the material. The detailed study of the spectra revealed the positions of $Sb3d_{5/2}$ and $Sn3d_{3/2}$ peaks located at 529.1 eV and 538.5 eV, respectively, identified as metallic Sb. The analysis of the

Te signal revealed the presence of metallic $Te3d_{5/2}$ and $Te3d_{3/2}$ peaks located at 573.2 and 583.6 eV, respectively.

For the both Bi₂Se₃ and Sb₂Te₃ ultrathin films, the detailed analysis of the XPS spectra revealed the presence of peaks related to the oxidized material. Shoulders in Bi4f spectra located at 159.3 eV and 164.6 eV (Fig. 2 a) corresponded to the oxidized surface of the Bi₂Se₃ film. A similar small intensity feature corresponding to oxidized Se is present in the spectrum of Se3d in the region of 59 eV (Fig. 2 b). For Sb₂Te₃, the oxidized Sb peaks located at 530.8 eV (3d_{5/} 2) and 540.2 eV (3d_{3/2}) with the additional peak located at 531.5 eV corresponding to O1s signal (Fig. 2 c), as well as oxidized Te3d_{5/2} and Te3d_{3/2} peaks located at 576.8 and 587.2 eV and Te monoxide located at 575.0 eV and 585.5 eV (Fig. 2 d) were identified. The presence of oxidation-related peaks in the XPS spectra of ultrathin Bi₂Se₃ and Sb₂Te₃ films is related to the susceptibility of these materials to rapid oxidation in air [24,25], as the synthesized ultrathin films were stored under ambient conditions for several months.

Chemical composition of the samples based on the measured XPS deconvoluted peak intensities has been evaluated to confirm the stoichiometry of the material (Table 1).

Raman spectroscopy of the deposited on different substrates ultrathin Bi_2Se_3 and Sb_2Te_3 films identified typical characteristic $E^2_{\rm g}$ and $A^2_{\rm 1g}$ vibration modes appearing at the peak positions of ~130 cm $^{-1}$ and ~173 cm $^{-1}$ corresponding to Bi_2Se_3 (Fig. 3 a), and ~117 cm $^{-1}$ and ~165 cm $^{-1}$ corresponding to Sb_2Te_3 (Fig. 3 b), which is consistent with the previous reports [26,27]. No significant deviations in the positions of the peak maxima were observed.

The X-ray diffraction (XRD) characterization of the ultrathin Bi₂Se₃ and Sb₂Te₃ films deposited on mica revealed that because of the very small thickness of the deposited Bi₂Se₃ films, the XRD diffraction peaks related to both Bi₂Se₃ and mica are overlapping, which greatly complicates spectra analysis (Fig. S2), while the XRD spectra of the ultrathin films deposited on quartz and graphene substrates allow to distinguish diffraction peaks related to the reflection planes of Bi₂Se₃ and Sb₂Te₃. Fig. 4 shows the representative XRD spectra of 6 nm thin Bi₂Se₃ film and 10 nm thin Sb₂Te₃

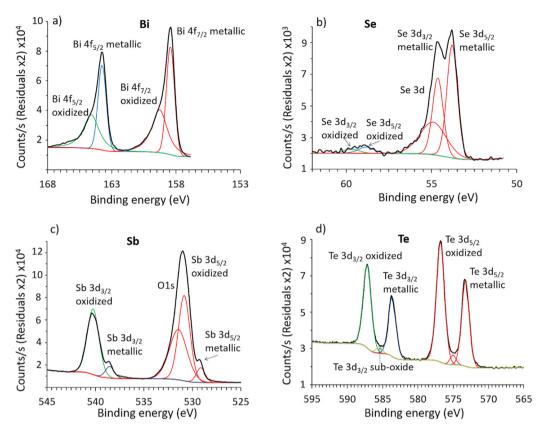


Fig. 2. Representative deconvoluted XPS spectra of a-b) 6 nm thin Bi₂Se₃ film and c-d) 5 nm thin Sb₂Te₃ film deposited on mica substrates. XPS, X-ray photoelectron spectroscopy.

Table 1Comparison of the chemical compositions of the Bi₂Se₃ and Sb₂Te₃ source materials and ultrathin films deposited on mica substrates revealed by the XPS measurements.

Chemical element	Bi ₂ Se ₃ source material at %	Bi ₂ Se ₃ 6 nm thin film on mica at %	Chemical element	Sb ₂ Te ₃ source material at %	Sb ₂ Te ₃ 5 nm thin film at %
Bi	37.1	36.7	Sb	41.6	44.9
Se	62.9	63.3	Te	58.4	55.1

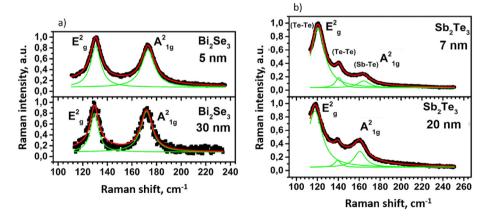
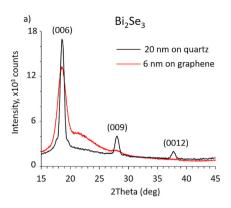


Fig. 3. Representative deconvoluted Raman spectra of a) 5 nm and 30 nm thin Bi₂Se₃ films deposited on quartz substrates, and b) 7 nm and 20 nm thin Sb₂Te₃ films deposited on mica substrates.

film deposited on graphene substrates. The XRD spectra of thicker (20 nm) Bi₂Se₃ and Sb₂Te₃ thin films are shown for the comparison.

The XRD analysis of the Bi₂Se₃ thin films spectra (Fig. 4 a) revealed the presence of peaks related the (003n) group and the

absence of peaks characteristic for polycrystalline Bi_2Se_3 (015) and (1010) reflection planes for both 6 nm and 20 nm thin films (ref. code 00-033-0214 [28]). This indicates well-oriented growth of Bi_2Se_3 , with crystallographic c-axis oriented perpendicularly to the



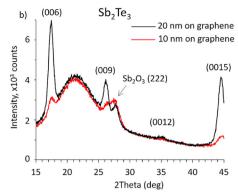


Fig. 4. Representative XRD spectra of a) Bi₂Se₃ and b) Sb₂Te₃ ultrathin films deposited on quartz and graphene substrates. XRD, X-ray diffraction.

substrate surface [20,29]. The intensity ratios for the peaks related to (006) and (009) reflection planes are 4.2 and 6.3 for the 20 nm thin film deposited on quartz and 6 nm thin film deposited on graphene, respectively, indicating a more uniform orientation of the ultrathin films grown on graphene in comparison with the ultrathin films grown on quartz, which is related to the epitaxial growth [20].

Similarly to the Bi_2Se_3 , the most of peaks presented in the XRD spectra of the Sb_2Te_3 thin films (Fig. 4 b) were related to the (003n) group (ref. code 01-071-0393, calculated from ICSD 002084 using POWD-12++). Besides these peaks, a peak located at 2Theta ~27.7 deg was observed in both Sb_2Te_3 spectra. This peak may be attributed to the (222) reflection plane of the Sb_2O_3 oxide (ref. code 01-075-1565, calculated from ICSD 031102 using POWD-12++) and confirms the XPS data, where slight excess of Sb presented in the oxidized form on the surface of the Sb_2Te_3 thin film was identified.

2.3. Electrical and thermoelectric characterization

Despite the same deposition method and similar growth orientation and chemical composition, the in-plane electrical and thermoelectric properties of the deposited under identical conditions on mica and quartz substrates Bi₂Se₃ and Sb₂Te₃ ultrathin films differ (Table 2).

As it is seen from Table 2, the ultrathin films deposited on quartz are few nm thicker and have higher Seebeck coefficients. However, because of the higher resistivity of these films, the resulting PF of these ultrathin films is lower than Bi₂Se₃ and Sb₂Te₃ ultrathin films deposited on mica. The reason for such a difference in properties may be mechanical and chemical defects of the quartz substrates, serving as multiple nucleation centers and resulting in formation of the film having more grain boundaries (boundaries between the coalesced nanoplates) than the ultrathin films deposited on mica, which is supported by the difference in intensity ratios of the XRD peaks, as well as by the previously reported structural investigation of the thicker films deposited on mica, quartz, and graphene [18,20]. While measurements of in-plane thermoelectric properties of ultrathin films deposited on graphene are challenging due to the

high electrical and thermal conductivity of graphene, it is expected that because of the epitaxial growth they will be similar to the ultrathin films deposited on mica. Multiple grain boundaries reduce electrical conductance of the ultrathin films, but at the same time may serve as energy filters for the charge carriers, which results in an increase of the Seebeck coefficient. The following sections of the manuscript are focused on the in-detail investigation of the properties of showing better thermoelectric performance Bi₂Se₃ and Sb₂Te₃ ultrathin films deposited on mica substrates.

2.4. Thickness-dependent electrical properties of ultrathin films deposited on mica

Fig. 5 a illustrates temperature dependence of the resistivity of the Bi_2Se_3 and Sb_2Te_3 ultrathin films. The Bi_2Se_3 ultrathin film with thicknesses 6 and 15 nm exhibit metallic behavior typical for thin Bi_2Se_3 nanostructures (Fig. 5 a) [30]. For these thin films, the higher decrease rate of resistivity of the latter possibly indicates higher concentration of contributing to the conductivity defects as Se vacancies [31]. In turn, the resistivity of the Bi_2Se_3 ultrathin film with thickness 5 nm reaches its minimal value at temperature 257 K and then increases with decreasing temperature.

Such upturn in the resistivity of 5 nm thin Bi₂Se₃ film is consistent with some previous reports related to the opening of an energy band gap in very thin Bi₂Se₃ samples due to the quantum confinement effects resulting in reduction of electrical conductance at low temperatures [32,33]. While the precise determination of the band gaps of the Bi₂Se₃ ultrathin films was outside the scope of this work, the relation of the band gap value to the thickness of the thin film can be roughly evaluated by the estimation of the Fermi energy and the assumption that in the non-doped semiconductor the band gap energy approximately equals to the double of it. The Fermi energy can be estimated from the experimentally determined Seebeck coefficient by the simplified formula $E_F \approx \pi^2 k_B^2 T/$ 3eS [34], where k_B is Boltzmann constant, T is the background temperature, e is the elementary charge, and S is the absolute value of the Seebeck coefficient. As it is seen from Fig. 6, the Fermi energy of the Bi₂Se₃ thin films tends to increase with the decrease of the

Table 2Representative comparison data between the properties of ultrathin Bi₂Se₃ and Sb₂Te₃ films deposited on mica and quartz substrates.

Compound	Substrate	Film thickness, nm	Seebeck coefficient, $\mu V \cdot K^{-1}$	Resistivity, Ω·m	PF, μ W·K ⁻² ·cm ⁻¹
Bi ₂ Se ₃	mica	6	-55	$1.23 \cdot 10^{-5}$	2.46
Bi ₂ Se ₃	quartz	8	-75	$1.7 \cdot 10^{-4}$	0.33
Sb ₂ Te ₃	mica	7	132	$0.8 \cdot 10^{-5}$	23
Sb ₂ Te ₃	quartz	10	154	$2.1 \cdot 10^{-5}$	11

PF, power factor.

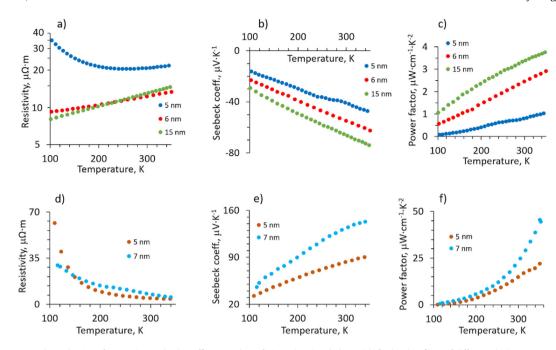


Fig. 5. Temperature dependencies of resistivity, Seebeck coefficient and PF of Bi₂Se₃ (a-c) and Sb₂Te₃ (d-f) ultrathin films of different thicknesses. PF, power factor.

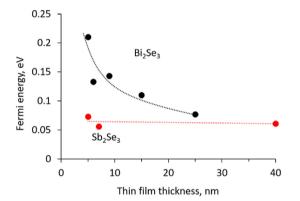


Fig. 6. Estimated values of the Fermi level for deposited on mica substrates Bi_2Se_3 and Sb_2Te_3 thin films of different thicknesses.

film thickness, proving the influence of the film thickness on the band gap size.

In contrast with the Bi_2Se_3 ultrathin films, the 5 nm and 7 nm thin Sb_2Te_3 showed the semiconductor type of conductivity (Fig. 5 d), with 5 nm thin film showing more pronounced temperature-dependent increase of resistivity and at a higher rate than 7 nm thin film. Accordingly to Hinsche et al. [14], such behavior of the resistivity may be attributed to significant contribution of surface states to the total conductivity of the thin films, whose hole concentration is in the order of $10^{18}~\rm cm^{-3}$. Increase of the Fermi level with the reduction of the film thickness was not observed for the 5–7 nm Sb_2Te_3 ultrathin films, which is in agreement with Osterhage et al. [10] which proposed opening of the hybridization gap in Sb_2Te_3 when its thickness is reduced below 4 nm.

2.5. The Seebeck coefficient and power factor of the ultrathin films deposited on mica

At 300 K, the Seebeck coefficient of the Bi_2Se_3 ultrathin films varied from $-67~\mu V~K^{-1}$ for 15 nm thick film to $-40~\mu V~K^{-1}$ for 5 nm

thick film (Fig. 5b). Negative values of Seebeck coefficients indicate n-type of conductivity of Bi₂Se₃. The PF calculated for 300 K varied from 3.4 μ W K⁻² cm⁻¹ for 15 nm thick film to 0.8 μ W K⁻² cm⁻¹ for 5 nm thick film. These values are comparable with the values obtained for the MBE-grown Bi_2Se_3 thin films (S = $-85 \mu V K^{-1}$ and $PF = 4 \mu W K^{-2} cm^{-1}$ for 15 nm thin film and $S = -64.6 \mu V K^{-1}$ and $PF = 1.4 \,\mu\text{W K}^{-2} \,\text{cm}^{-1}$ for 5 nm thin film) [13] and indicate usability of our PVD method for obtaining of high-quality Bi₂Se₃ ultrathin films. For the Sb₂Te₃ ultrathin films, Seebeck coefficients were positive, indicating p-type conductivity as is expected for this material. 7 nm thin film showed significantly higher Seebeck coefficient than the 5 nm thin film (S = 130 μ V K⁻¹ vs. S = 82 μ V K⁻¹). The calculated PF for 7 nm and 5 nm thin Sb₂Te₃ films were 23 μ W K⁻² cm⁻¹ and 15 μ W K⁻² cm⁻¹, respectively. These values are comparable with ones reported for ~1 µm thick Sb₂Te₃ thin films grown by MBE (S = 130 μ V K⁻¹, PF = 29 μ W K⁻² cm⁻¹) [35], co-evaporation technique (S = 160 μ V K⁻¹, PF = 20 μ W K⁻² cm⁻¹) [36], and bulk Sb₂Te₃ grown by the Bridgman method (S = 79 μ V K⁻¹, PF = 30 μ W K⁻² cm⁻¹) [35]. There is a lack of experimental data on thermoelectrical properties on the ultrathin (below 30 nm) Sb₂Te₃ films: however, accordingly to the theoretical studies, the significant difference in Seebeck coefficients of 5 nm and 7 nm thin Sb₂Te₃ films may be related to the larger contribution of topological conductive surface states in thinner films [14]. For easier reading, Table 3 summarizes Seebeck coefficient and PF of the thin films obtained in this work and reported by other research groups, including data for bulk Bi₂Se₃ and Sb₂Te₃ materials.

Temperature dependence of experimentally determined Seebeck coefficients and estimated PF of Bi₂Se₃ and Sb₂Te₃ ultrathin films of different thicknesses are shown in Fig. 5 b,c,e,f. All Bi₂Se₃ films showed quasilinear decrease of the Seebeck coefficient and PF with the decrease of temperature (Fig. 5 b,c), which indicates prevailing degenerate electron state [13]. Determined from the slopes of the PF curves decrease rates showed clear dependence of PF on the ultrathin films thickness, and ranged from 0.0113 μ W cm⁻¹ K⁻² per K for 15 nm thin film down to 0.0038 μ W cm⁻¹ K⁻² per K for 5 nm thin film. Lower PF decrease rates of thinner Bi₂Se₃ films indicate their better applicability for

Table 3Comparison between the Seebeck coefficients and PF of Bi₂Se₃ and Sb₂Te₃ thin films obtained in this work and by other research groups.

Sample	Seebeck coefficient, $\mu V \cdot K^{-1}$	PF, μW⋅K ⁻² ⋅cm ⁻¹	Research group	
5 nm thin Bi ₂ Se ₃ film	-40	0.8	this work	
6 nm thin Bi ₂ Se ₃ film	-55	2.45	this work	
15 nm thin Bi ₂ Se ₃ film	-67	3.5	this work	
MBE-grown 5 nm thin Bi ₂ Se ₃ film	-64.6	1.4	Guo et al. [13]	
MBE-grown 15 nm thin Bi ₂ Se ₃ film	-85	4.0	Guo et al. [13]	
Bulk Bi ₂ Se ₃ (Bridgman method)	-59	9.4	Navrátil et al. [37]	
5 nm thin Sb ₂ Te ₃ film	82	15	this work	
7 nm thin Sb ₂ Te ₃ film	130	23	this work	
MBE-grown 1 μm thin Sb ₂ Te ₃ film	130	29	Peranio et al. [35]	
Co-evaporated 1 µm thin Sb ₂ Te ₃ film	160	20	Huang et al. [36]	
Bulk Sb ₂ Te ₃ (Bridgman method)	79	30	Peranio et al. [35]	

MBE, molecular beam epitaxy.

wide-range temperature applications. Observed for the considered temperature range systematic decrease of the Seebeck coefficient and PF with the decrease of Bi_2Se_3 ultrathin film thickness is consistent with the previous report on thermoelectrical properties of Bi_2Se_3 MBE thin films with thicknesses 5–30 nm [13]. It can be attributed to the increase of the contribution of Dirac-like topological surface states, whose linear dispersion is not optimal for the increase of the Seebeck coefficient [13]. The possible measures for improvement of the power factor of ultrathin films may include further decrease of the film thickness below 5 nm to completely gap the surface states, as well as tuning of the Fermi level by doping. Temperature dependencies of Seebeck coefficient of Sb_2Te_3 ultrathin films showed slight deviations from linear behavior, but

their PF exhibited power-law increase (especially for the 7 nm thin Sb_2Te_3 film) with the increase of temperature (Fig. 5 e,f). The nonlinear behavior of Seebeck coefficient and PF of ultrathin Sb_2Te_3 films is consistent with the remarkable contribution of topological surface states of the nanoscale Sb_2Te_3 with charge carrier concentrations in the order $10^{18}-10^{19}$ [14]. However, further experimental work is required for clarification of the observed dependencies. Significant increase of the PF of the 7 nm thin Sb_2Te_3 film above the room temperature makes it attractive for application in thermoelectrical devices for domestic waste heat conversion.

Showing the best PF Bi_2Se_3 and Sb_2Te_3 ultrathin films of 15 nm and 7 nm thickness, respectively, were tested in a prototype of two-leg TEG as described in the Experimental section.

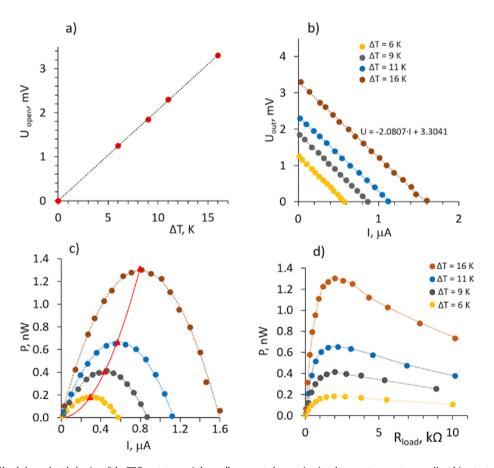


Fig. 7. Characterization of load-dependent behavior of the TEG prototype: a) thermally generated open circuit voltage vs. temperature gradient b) output voltage vs. output current; c) output power vs. output current; red triangles represent maximal power for each temperature gradient; d) output power vs. external load.

Performance of the TEG prototype was characterized by measuring the load-dependence behavior [38]. The open circuit output voltage generated by the TEG showed linear dependence on the temperature gradient (Fig. 7 a) that can be described by function U_{out} [mV] = $0.2 \cdot \Delta T[K]$. This result is roughly 2 times lower in comparison with the $U_{\text{open}}(\Delta T)$ function obtained for commercial thermoelectric module TEC1-12707 ($U_{open}[mV] = 53.17 \cdot \Delta T$ for 127 leg couples, which equals $U_{open}[mV] = 0.42 \cdot \Delta T$ per one leg pair) [38]. Most likely, the poorer performance of the demonstrated prototype in comparison with the commercial device is related to the heat dissipation at the ultrathin film—mica substrate interface and in the electrical contacts to the ultrathin films, which should be optimized in future. The output voltage vs. output current of the TEG at different external loads and temperature gradients is shown in Fig. 7 b. Dependencies of output voltage vs. output current are linear and of equal slope, proving stable operation of the device and allowing to extract the thermopower and internal resistance of the TEG from the linear fittings (Fig. 7 b), as well as to calculate the output power vs. output current (Fig. 7 c) and vs. the external load (Fig. 7 d) (Table 4). It was found that the maximal output power produced by the TEG prototype shows power-law increase with the increase of the temperature gradient (Fig. 7 c), which makes these devices attractive for applications involving low temperature gradients, for example, for domestic waste heat conversion. As the maximal power generated by the two-leg TEG prototype was in the range of nW for the nanosized cross-section, the comparison of its performance with the performance of the commercially available thermoelectric module was performed per cross-section unit. The maximal powers per cross-section area generated by the two-leg TEG prototype and 127 leg couple TEC1-12707 [38] were found to be 1.2 mW cm⁻² and 6.25 mW cm⁻², respectively, indicating a high potential of the demonstrated TEG prototype for further development and application in domestic waste heat conversion

As it can be seen from Table 4, the main drawback of the developed two-leg TEG prototype is high internal resistance in comparison with the bulk materials because of the very high length-to-cross-section ratios of the ultrathin films. This puts limitation to the resistance of the external devices as it is well known that power transfer between a voltage source and an external load is at its most efficient when the resistance of the load matches the internal resistance of the voltage source. However, such TEGs may find application for powering of potential devices having the resistance in the range 1–10 $k\Omega$ as their operation at such external loads is quite stable (Fig. 7 d). Comparison of the properties (measured at room temperature) of ultrathin films used for development of two-leg TEG prototype with the properties of the thin films used for development of the superlattice-based thin film thermoelectric modules, capable of producing a cooling flux of double that of the current state-of-the-art value [39]; however, not tested under different external loads is shown in Table 5.

As it is seen from the data in Table 5 and also supported by the research performed by other groups [40], the improvement of the thermoelectric efficiency of thin films may be achieved by consecutive deposition of different materials having the same type

Table 4The main parameters of the ultrathin films—based two-leg TEG prototype.

Temperature gradient, K	Thermopower, μV/K	Internal resistance, Ω	Maximal output power, nW	Optimal load resistance, Ω
6	204.9	2195	0.184	2188
9	203.6	2158	0.406	2136
11	205.5	2061	0.666	2057
16	200.6	2081	1.316	2087

Table 5Comparison between properties of thin films synthesized in this work and reported by the other group.

Type of the thin film	Electrical resistivity, $\mu\Omega\cdot m$	Seebeck coefficient, $\mu V \cdot K^{-1}$	PF, μW·K ⁻² ·cm ⁻	Reference
p-type Bi ₂ Te ₃ / Sb ₂ Te ₃	10.2	238	55	Bulman et al. [39]
p-type Sb ₂ Te ₃	7.5	132	23	This work
n-type δ-doped Bi ₂ Te _{3-x} Se _x	13.7	-276	56	Bulman et al. [39]
n-type Bi ₂ Se ₃	13	-67	3.5	This work

of conductivity and similar crystal lattices, which is in the scope of the following work of our group.

3. Conclusions

In summary, the PVD method for obtaining of Bi₂Se₃ and Sb₂Te₃ ultrathin films of thicknesses 5–10–15 nm on different substrates as mica, fused quartz, and monolayer graphene has been developed. 5–7 nm thin Sb₂Te₃ films were synthesized for the first time. Deposited on mica 5-15 nm thin Bi₂Se₃ films and 5-7 nm thin Sb₂Te₃ films showed thermoelectrical properties comparable with MBE-grown thin films. 5 nm thin Bi₂Se₃ film showed pronounced unusual upturn in resistance at temperatures below 257 K. These results are consistent with recent theoretical research gapping of topological surface states in Bi₂Se₃ thin films with thicknesses below 6 nm, which may potentially result in enhancement of thermoelectrical efficiency of this material. However, the observed systematic decrease of the Seebeck coefficient and power factor with the decrease of thickness of the Bi₂Se₃ ultrathin films indicates necessity of additional downsizing measures for effective enhancement of thermoelectric properties of this material. In turn, 7 nm thin Sb₂Te₃ film showed significant thermoelectric properties. which are comparable with 1 µm-thick MBE-grown film. Performance of 15 nm thin Bi₂Se₃ film and 7 nm thin Sb₂Te₃ film was tested in the two-leg prototype of the TEG. It showed stable operation under different external loads and temperature gradients and performance comparable with the commercial thermoelectric module; however, because of the high internal resistance it was found useful for powering devices with resistances in the range 1–10 k Ω . Potential routes for the improvement of the thermoelectric performance of the ultrathin films-based TEGs include replacement of the single ultrathin film with multilayered structures, the use of substrates with low thermal conductivity, and optimization of the electrical contacts to ultrathin films.

4. Materials and methods

4.1. Deposition of ultrathin films

The ultrathin films were deposited using a single-zone quartz tube furnace GCL-1100X (MTI Corp.). Bismuth (Bi) (\geq 99.99%, CAS: 7440-69-9, Sigma Aldrich), selenium (Se) (99+, CAS: 7782-49-2, Emsure) and antimony telluride (Sb₂Te₃) (99.999%, CAS: 1327-50-0, Alfa Aesar) powders were used as source materials. Evaporation temperatures were 500 °C for Bi, 480 °C for Se, and 415–420 °C for Sb₂Te₃. The substrate – freshly cleaved natural mica sheets – of thickness ~60 μ m (Agar Scientific) or fused quartz 2 cm \times 2 cm x 1 mm (Agar Scientific) or grown by chemical vapor deposition and transferred to the quartz substrate monolayer graphene [29] was placed downstream from the source materials located in the central area of the furnace tube. After the placement of the source materials and the substrate, the furnace tube was ventilated with

the nitrogen flow and pumped down to the base pressure of 150 mTorr, which was kept constant during the deposition process. The heating rate of the source materials was 48 °C/min. The maximal substrate temperature during the deposition process was 400 °C for deposition of Bi₂Se₃ selenide and 220 °C for deposition of Sb₂Te₃. The furnace was held at the maximal source materials evaporation temperature for 1–3 min for deposition of Bi₂Se₃ of different thicknesses, followed by natural cooling down to the room temperature. For obtaining of the ultrathin Sb₂Te₃ films, after reaching the maximal temperature in the center of the furnace hot zone, the substrate and the source material were rapidly cooled down by extracting the quartz furnace tube out of the hot zone of the furnace and simultaneous filling it with nitrogen up to the atmospheric pressure.

4.2. Characterization

Morphology, thickness, and structure of the deposited ultrathin films were inspected using the field emission SEM Hitachi S-4800, AFM Asylum Research MFP-3D, and X-ray diffraction spectroscopy (powder diffractometer X'PERT MRD with Cu K_{α} radiation source). XPS analyses were carried out using the ThermoFisher ESCALAB Xi⁺ instrument using a monochromatic Al Ka X-ray source. The instrument binding-energy scale was calibrated to give a binding energy at 932.6 eV for Cu 2p_{3/2} line of freshly etched metallic copper. The charge compensation system was used on all nonconductive specimen samples. The surface of each sample was irradiated with a flood of electrons to produce a nearly neutral surface charge. The spectra were recorded by using an X-ray beam size 900x10 µm, a pass energy of 20 eV, and step size 0.1 eV. Data from all materials have been referenced using the main signal of the carbon 1s spectrum assigned to occur at 285.0 eV. The carbon 1s spectrum was collected using high energy resolution settings. Raman spectra were recorded at the room temperature using scanning laser confocal micro-Raman spectrometer Confotec NR 500 with 785 nm excitation wavelength, 0.86 mW power, and 100x objective. The accumulation time of the signal for each spectrum was 10 s. The laser beam diameter was about 600 nm. Raman spectra were registered using 1800 l/mm grating, and spectral resolution ~0.2 cm⁻¹. Throughout these measurements, the edge filter cutting at ~85 cm⁻¹ was used.

Electrical and thermoelectric characterization: Resistance and Seebeck coefficients of the ultrathin films were performed using thermal transport option of Physical Property Measurement System DynaCool9T (Quantum Design) in 4-point configuration for elimination of the influence of contact resistance. For determination of the Seebeck coefficient, the temperature difference between the sides of the ultrathin films did not exceed 3% from the measurement temperature.

4.3. Fabrication and testing of a TEG prototype

For the fabrication of the TEG prototype consisting of one p-type and one n-type leg, the 5 mm \times 15 mm large stripes of Bi₂Se₃ and Sb₂Te₃ ultrathin films deposited on mica were used. 5 mm \times 5 mm gold contacts (thickness ~ 100 nm) were deposited over the sides of the thermoelectric stripes using the thermal deposition method (SIDRABE SAF EM sputtering vacuum system), leaving 5 mm \times 5 mm of the ultrathin film between the electrodes free. Copper wires were attached to the gold electrodes using electrically conductive adhesive (EPO-TEK®H20E). Thermally generated by the TEG prototype voltage was measured using a homemade device for thermoelectrical measurements [20] under temperature gradients

from 6 up to 12 K, as open circuit and under different (from 1 k Ω up to 100 k Ω) external loads.

Credit author statement

Jana Andzane: Investigation, Formal analysis, Visualization, Writing — original draft **Andrei Felsharuk:** Investigation, Methodology, Validation **Anatolijs Sarakovskis:** Investigation, Formal analysis, Visualization, Writing — original draft **Uldis Malinovskis:** Investigation, Formal analysis **E. Kauranens:** Investigation, Formal analysis, Visualization **Mikhael Bechelany:** Investigation, Formal analysis **Kiryl A. Niherysh:** Investigation, Formal analysis, Visualization **Ivan V. Komissarov:** Conceptualization, Supervision **Donats Erts:** Conceptualization, Methodology, Supervision, Writing — review & editing

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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The raw/processed data required to reproduce these findings cannot be shared at this time as the data also form a part of an ongoing study.

Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.mtener.2020.100587.

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