

Synthesis and characterization of GaN/ReS₂, ZnS/ReS₂ and ZnO/ReS₂ core/shell nanowire heterostructures

Edgars Butanovs^{1,*}, Alexei Kuzmin¹, Sergei Piskunov¹, Krisjanis Smits¹, Aleksandr Kalinko², Boris Polyakov¹

¹ Institute of Solid State Physics, University of Latvia, Kengaraga Street 8, Riga, Latvia, LV-1063

² Department Chemie, Naturwissenschaftliche Fakultät, Universität Paderborn, Warburger Strasse 100, 33098 Paderborn, Germany

*e-mail: edgars.butanovs@cfi.lu.lv

Abstract

Layered 2D van der Waals (vdW) materials such as graphene and transition metal dichalcogenides have recently gained a great deal of scientific attention due to their unique properties and prospective applications in various fields such as electronics and optoelectronics, sensors and energy. As a direct bandgap semiconductor in both bulk and monolayer forms, ReS₂ stands out for its unique distorted octahedral structure that results in distinctive anisotropic physical properties; however, only a few scalable synthesis methods for few-layer ReS₂ have been proposed thus far. Here, the growth of high-quality few-layer ReS₂ is demonstrated via sulfurization of a pre-deposited rhenium oxide coating on different semiconductor material nanowires (GaN, ZnS, ZnO). As-produced *core-shell* heterostructures were characterized by X-ray diffraction, scanning and transmission electron microscopy, micro-Raman spectroscopy and X-ray absorption spectroscopy. Experimental characterizations were supported by total energy calculations of the electronic structure of ReS₂ nanosheets and GaN, ZnS, and ZnO substrates. Our results demonstrate the potential of using nanowires as a template for the growth of layered vdW materials to create novel *core-shell* heterostructures for energy applications involving photocatalytic and electrocatalytic hydrogen evolution.

Keywords: *ReS₂; layered materials; core-shell nanowire; electron microscopy; X-ray diffraction; Raman spectroscopy; X-ray absorption spectroscopy*

1. Introduction

Layered 2D van der Waals (vdW) materials such as graphene and transition metal dichalcogenides (TMDs) have recently gained a great deal of attention due to their unique properties when the bulk material is reduced to a monolayer [1–3], the wide variety of available materials [4] and their prospective applications in various fields, such as electronics and optoelectronics [5,6], sensors [7] and energy [8]. Currently, one of the main limitations on the use of these materials in practical devices is the growth of uniform, high-quality large-area films with controllable thicknesses [4,5]. Since the layers are bound together by weak vdW interactions, 2D materials to some extent do not need to satisfy the conventional semiconductor lattice matching requirement to grow well-crystallized epitaxial layers of different materials on top of each other [9,10]. Hence, the growth of high-quality 2D heterostructures of some of the most popular layered materials on various substrates has already been demonstrated [11,12], while techniques for growing more recently explored layered materials such as HfS₂ and ReS₂ are still being established [5,13].

The energy-related applications of TMD materials have recently attracted significant interest. For example, TMD materials can be used in supercapacitors, hydrogen evolution reactions and Li-ion battery (LIB) applications [14,15]. However, the bulk TMDs used in LIBs demonstrate rapid capacity fading. The engineering of novel 3D TMD-based hybrids is a potential approach to solving this issue and to improving the electrochemical performance of TMD nanomaterials [16]. One

promising option is use of *core-shell* nanowire (NW) array architecture, in which the “core” is used as a high-mobility charge carrier channel, while the TMD-based “shell” is used for ion accumulation or electrochemical ion intercalation. This type of heterostructure design could also be used for efficient photocatalytic and electrocatalytic hydrogen evolution reactions. Very recently, the synthesis of novel TiO₂-ReS₂ *core-shell* heterostructures was reported by Wang *et al.* [17], and these have demonstrated very stable performance as a sodium-ion storage materials.

NWs are one-dimensional (1D) nanomaterials that exhibit promising properties e.g. for integration in electronic, optoelectronic and photonic functional devices [18,19]; moreover, their characteristics can be further improved by modification of their surfaces [20]. It has been shown that the surface of a NW has significantly smaller restriction on the lattice mismatch of the materials and the stresses caused at the interface, in contrast to the conventional thin films growth [18,20,21], thus offering new possibilities of engineering novel hybrid *core-shell* nanostructures. The growth of various common 2D materials such as MoS₂ and WS₂ on different semiconductor NWs has been previously demonstrated [22–24], thus proving the possibility of using NWs as a prospective template for the growth of high-quality layered vdW materials, and even of achieving improved physical properties [23,25].

Similarly to other TMDs, ReS₂ consists of covalently bonded S-Re-S atomic planes, and these adjacent layers are coupled by weak vdW forces to form a bulk crystal [13]. It is important to note that unlike other 2D materials with an indirect-to-direct bandgap transition in monolayers, ReS₂ exhibits much weaker interlayer coupling, and therefore remains a direct bandgap ($E_g \sim 1.5 - 1.6$ eV) semiconductor in the bulk, few-layer and monolayer forms [26]. It possesses a unique distorted octahedral (1T) structure [13] that results in anisotropic physical properties along the in-plane directions

[27,28]. This anisotropy leads to distinctive optical properties [29], anisotropic electrical transport [28,30], thermal conductivity [31] and piezoresistive effect [32], which may be useful for emerging applications in electronics and optoelectronics [27] such as polarized light detection [33]. Other demonstrated applications of 2D ReS₂ include field-effect transistors [26,34,35], photodetectors [36–38], digital inverters [39], tunnel diodes [40], and photocatalytic and electrocatalytic hydrogen evolution (water splitting) [41].

Large-scale growth of uniform crystalline ReS₂ layers with controllable thickness is paramount for any practical applications. Mechanical exfoliation of a bulk crystal is commonly used in fundamental research [28,29,34,39] to obtain the highest quality crystals; however, this produces small mono- or few-layer flakes with uncontrollable position on the substrate, thereby making fabrication of devices more complicated. Chemical vapour transport (CVT) has been recently reported as a viable method of growing uniform ReS₂ films. He *et al.* [42] used metallic Re as a precursor to obtain high-quality ReS₂ flakes; however, due to the high melting point of Re (3180°C), which gives rise to a low vapour pressure in the CVT system, this approach inevitably provides a low yield and a small growth rate of ReS₂. Several groups have demonstrated the growth of large-area ReS₂ at relatively low temperatures using ammonium perrhenate (NH₄ReO₄) [43], ReO₃ [36,44] or Re₂O₇ [45] as the rhenium precursor. The use of NH₄ReO₄ typically leads to poor crystal quality, mainly due to the high oxidation state of Re (+7) and unwanted byproducts [13]. Furthermore, Cui *et al.* [46] demonstrated that the synthesis temperature of ReS₂ can be lowered to 430°C by forming a Re-Te eutectoid, thus producing large-area, high-crystallinity and uniform monolayers on mica with the tellurium-assisted CVT method. Since ReS₂ exhibits strong interlayer decoupling, out-of-plane growth is predominant on many substrates,

including the widely used SiO₂/Si, leading to the growth of wrinkled low-quality polycrystalline films or thick flower-like ReS₂ via a typical CVT process [45]. Thus, ReS₂ is commonly grown on mica substrates due to its low surface energy [45,46], and on other 2D vdW materials such as graphene [44]. However, a crucial disadvantage of using mica is the need for a highly corrosive hydrofluoric acid for etching in order to delaminate the ReS₂ film from the substrate for further application [13]. Alternatively, sulfurization of a pre-deposited transition metal or transition metal oxide film has been successfully utilized for numerous TMDs to obtain high-quality mono- and few-layer films [47–50]. Urakami *et al.* [51] demonstrated the formation of ReS₂ films via the sulfurization of e-beam-deposited Re films on sapphire substrates, although a high temperature of 1100°C was necessary. To the best of our knowledge, no previous reports on the sulfurization of pre-deposited rhenium oxide films to obtain few-layer ReS₂ have been published.

In this study, we demonstrated the growth of few-layer ReS₂ on different semiconductor material NWs (GaN, ZnS, ZnO) via the sulfurization of a pre-deposited rhenium oxide coating, and investigated the degree of ReS₂ crystallinity at different sulfurization temperatures. This two-step process allowed us to obtain high crystalline quality ReS₂ on different underlying materials in a medium temperature range. The results demonstrate the potential of using NWs as a template for growing layered vdW materials to produce novel hybrid *core-shell* nanostructures and 3D architectures. Furthermore, according to predictions made from our first principles calculations, these *core-shell* heterostructures could be used in efficient hydrogen evolution reactions [14].

2. Experimental section

2.1. Nanowire synthesis

GaN, ZnS and ZnO NWs were synthesised using an adapted atmospheric pressure chemical vapour transport method in a horizontal quartz tube reactor [24,52,53]. In all cases, the NWs were grown on oxidized silicon wafers SiO₂/Si(100) (*Semiconductor Wafer, Inc.*) coated with spherical Au nanoparticles (*Smart materials*, water suspension, 100 nm diameter) which were used as a catalyst for the vapour-liquid-solid (VLS) mechanism.

GaN NWs: 3 g metallic Ga (99.999%, *Alfa Aesar*) was loaded into a ceramic boat and placed in the centre of the quartz tube, and Au/Si substrates were placed downstream in a lower-temperature region. The reactor was heated to 920°C under a flow of carrier gas mixture Ar/H₂-35%, and gaseous NH₃ flow in a 1:1 ratio to the carrier gas was then introduced and maintained for 30 minutes to enable the gas-phase reaction and NW growth, followed by natural cooling to room temperature under Ar/H₂ flow. As a result, GaN NWs of length 5 – 10 μm were produced on the SiO₂/Si substrate (see Fig. S1(a,b))

ZnS NWs: 0.4 g ZnS powder (>97%, *Sigma Aldrich*) was thermally evaporated in a quartz tube at a temperature of 950°C for 30 minutes, followed by natural cooling. The vapour was carried downstream to the Au/Si substrate by N₂ gas to grow ZnS NWs of length 20 – 100 μm (see Fig. S1(c,d)).

ZnO NWs: A 1:1 mixture of ZnO and carbon powders was loaded into a ceramic boat in the centre of the quartz tube at 900°C, and the vapour was transported downstream to the Au/Si substrate in a lower-temperature region using N₂ as a carrier gas. The temperature during growth was held constant for 90 minutes, followed by natural cooling to room temperature. As a result, ZnO NWs of length 10 – 50 μm were obtained on the SiO₂/Si substrate (see Fig. S1(e,f)).

2.2. ReS₂ synthesis

A simple two-step process was used to produce the ReS₂ shell surrounding the pure NW core: (1) a rhenium oxide (ReO_x) coating was sputter-deposited on top of the NW substrate followed by (2) sulfurization at elevated temperatures. First, an amorphous nonstoichiometric ReO_x coating was deposited via reactive DC magnetron sputtering of a metallic rhenium (Re) target in a mixed Ar/O₂ atmosphere (1 minute at 100W DC power) using a *Sidrabe SAF25/50* multifunctional cluster tool. Then, in order to convert rhenium oxide to ReS₂, as-prepared ReO_x samples were annealed for 20 minutes in a quartz tube reactor in a sulfur-rich atmosphere, which was created by placing sulfur powder upstream of the sample at 250°C and using N₂ as a carrier gas. Different temperatures in the range 450-800°C were used to study the degree of crystallinity of as-grown ReS₂. ReS₂ thin films on SiO₂/Si(100), and GaN-ReS₂, ZnS-ReS₂, ZnO-ReS₂ *core-shell* NWs were prepared.

As a reference sample, ReS₂ microcrystals were grown using the vapour transport method. In brief, a quartz ampoule was loaded with elemental precursor powders (Re and S) in stoichiometric proportions and a small amount of transport agent (iodine) powder was added. The quartz ampoule was then evacuated and sealed. A small temperature gradient was maintained between the ends of the ampoule ($T_{\text{high}} \sim 1100^\circ\text{C}$ and $T_{\text{low}} \sim 1050^\circ\text{C}$) for several days, and the growth of micrometre-sized ReS₂ monocrystals occurred at the low-temperature end.

2.3. Core-shell nanowire characterization

The morphology of the as-prepared *core-shell* NWs was characterized by scanning electron microscopy (SEM, Lyra, Tescan), whereas the inner crystalline

structure was revealed by transmission electron microscopy (TEM, Tecnai GF20, FEI) operating at an accelerating voltage of 200 kV. Energy-dispersive X-ray spectroscopy (EDX) was used in both SEM and TEM to study the distribution of elements in the NWs. A selected-area electron diffraction (SAED) analysis was performed using CrysTBox software [54].

The phase composition of the samples was studied with X-ray diffraction (XRD) using a Rigaku MiniFlex 600 X-ray powder diffractometer with Bragg-Brentano θ - 2θ geometry and a 600W Cu anode (Cu K α radiation, $\lambda = 1.5406 \text{ \AA}$) X-ray tube.

Micro-Raman spectroscopy measurements were performed using a TriVista 777 confocal Raman system (Princeton Instruments, 750 mm focal length, 1800 lines/mm grating) equipped with an upright Olympus microscope with Olympus UIS2 MPlanN 100x/0.90 objective, a continuous-wave single-frequency diode-pumped laser Cobolt Samba 150 ($\lambda=532 \text{ nm}$) and an Andor iDus DV420A-OE CCD camera.

X-ray absorption spectroscopy (XAS) measurements were performed at the HASYLAB/DESY PETRA-III P64 beamline [55]. The X-ray beam from the undulator was monochromatized with a Si(311) double-crystal monochromator, and the X-ray absorption near edge (XANES) spectra were recorded at the Re L₃-edge and Ga K-edge in fluorescence mode with a Passivated Implanted Planar Silicon (PIPS) detector (Canberra). All XAS measurements were made at room temperature.

2.4. Computational details

We carried out total energy first principles calculations on [0001]-oriented mono- (ML), bi- (2ML), and three-layered (3ML) ReS₂ nanosheets, as well as 24-layer thick [1-100] oriented slabs, in order to mimic the surfaces of GaN, ZnO and ZnS NWs.

This was achieved by means of a HSE06 hybrid exchange-correlational functional [56] within density functional theory (DFT), as implemented in the computer code CRYSTAL17 [57]. Localized Gaussian-type functions (GTFs) in the form of atom-centred basis sets (BSs) were used for the expansion of periodic crystalline orbitals for Zn, S, and O, in the same way as in our recent study [20]. For Ga and N, full electron Triple-Zeta Valence BSs with polarization functions were adopted from [57], while effective-core pseudopotential BSs were used for Re [57]. The reliability of the chosen theoretical method was proven by calculations of band gap energy (δ) for the bulk phase of GaN, ZnS, ZnO, and ReS₂ crystals. Calculated band gaps for all of the materials under study showed good agreement with the experimentally observed results (see Fig. S6). To provide a balanced summation for both the direct and reciprocal lattices, reciprocal space integration was performed by sampling the Brillouin zone (BZ) with a $6 \times 6 \times 1$ Pack-Monkhorst mesh [58], which resulted in a total of 20 k-points evenly distributed over the BZs. In each calculation of the fixed crystalline geometry, convergence was reached only when the total energy differed by less than 10^{-7} a. u. in two successive cycles of the self-consistent field (SCF) procedure [57]. Full geometry optimization was performed for all of the nanostructures considered in this study.

3. Results and discussion

NWs can be used as a template for the growth of layered vdW materials in order to conveniently study their inner crystalline structure via TEM. Fig. 1 shows TEM images of as-grown GaN-ReS₂ (a,b), ZnS-ReS₂ (c,d) and ZnO-ReS₂ (e,f) *core-shell* NWs prepared at 800°C. The lower-magnification images show that GaN and ZnS NWs maintain their straight shape; however, ZnO NWs undergo a transition to an irregular shape, indicating the occurrence of a recrystallization process. The higher-

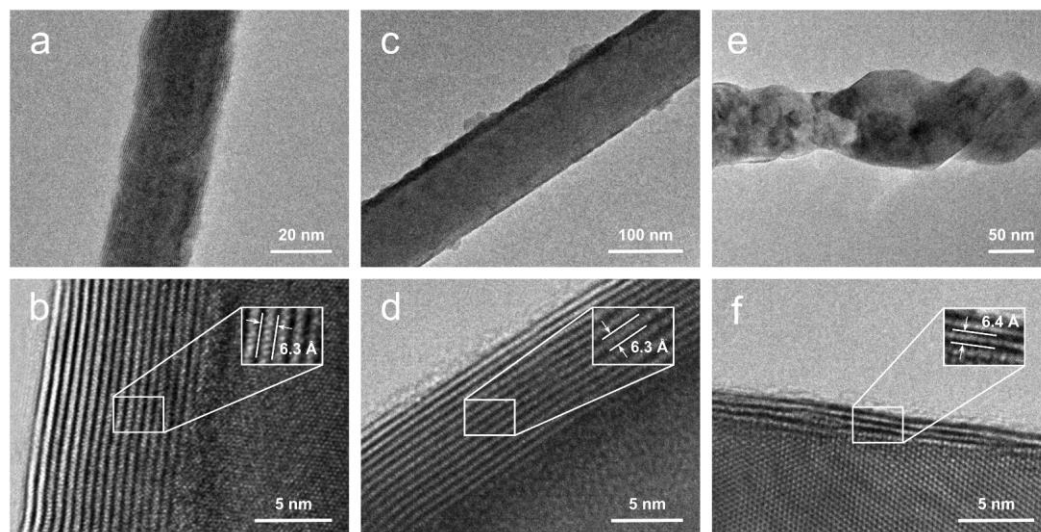


Figure 1. Transmission electron microscope images at different magnifications for as-grown (a,b) GaN-ReS₂, (c,d) ZnS-ReS₂, and (e,f) ZnO-ReS₂ *core-shell* nanowires, prepared at 800°C. The insets show the measured atomic interlayer distances between ReS₂ layers.

magnification TEM images reveal the layered structure of the ReS₂ shell (which can be seen as parallel black and white lines along the NW surface) with a measured interlayer distance of around 6.3-6.4 Å for all samples, which closely matches the lattice constant $a = 6.45$ Å of triclinic ReS₂ [59]. The typical thickness of the ReS₂ shell varied from 2-10 monolayers (each consisting of S-Re-S atomic planes), and the shell was uniformly distributed over the length of the NWs with only slight variations in thickness. Furthermore, the single-crystalline nature of the core NWs is visible (in NWs with diameter less than 80 nm) indicating the high crystalline quality of the as-prepared *core-shell* heterostructures, as confirmed by SAED measurements. A detailed description of the SAED analysis is given in Fig. S2. Dominating (most intensive) diffraction spots typically belong to the NW core material, and are less intensive for the ReS₂ shell. The phases and crystallographic orientation were identified through a separate analysis of each pattern. The symmetric orientation of ReS₂ reflexes correlated with ZnO reflexes

might indicate the epitaxial growth of ReS₂, while the SAED patterns of ReS₂ on ZnS and GaN do not exhibit epitaxial growth.

SEM images of the as-grown nanostructures (see Fig. 2) show that the length of the NWs was preserved, matching their respective pure NWs before the high-temperature treatment, and the density of the NW arrays was maintained. Similarly to TEM measurements, the morphology change of ZnO-ReS₂ NWs upon heating at 800°C can also be observed by SEM.

To study the distribution of elements in the NWs, EDX measurements were performed (see Fig. 3). First, EDX elemental mapping along the length of the NWs was carried out via SEM to demonstrate the uniform distribution of ReS₂ axially over the underlying NWs. Next, cross-sectional EDX line scans were performed radially across the NWs using TEM, in order to prove the *core-shell* distribution of the elements. The elemental line scans show a localized core and shell with distinct interfaces in all studied NWs, a result that is consistent with the TEM images. It should be noted that the EDX intensity is usually not symmetric on both sides of the NW due to the sideways placement of the EDX detector in the electron microscope chamber.

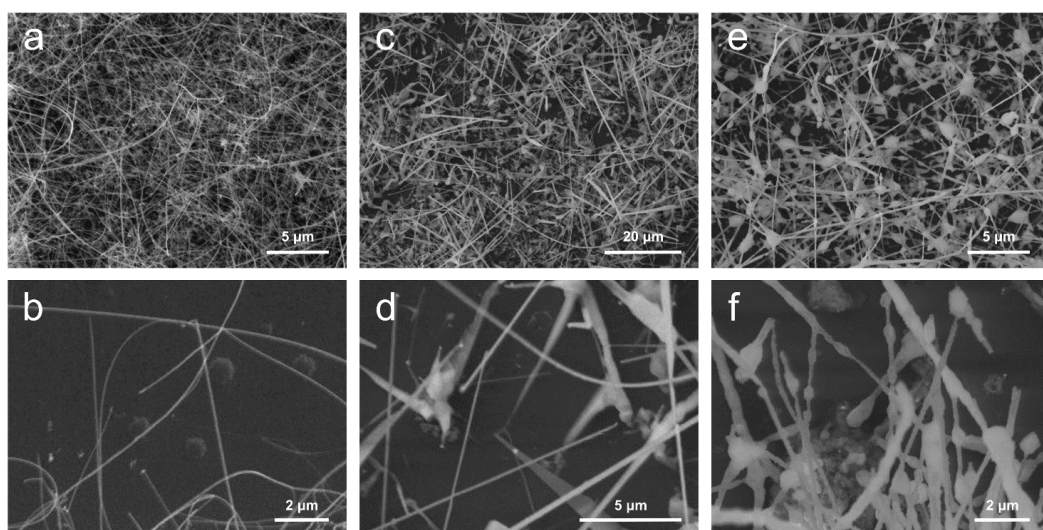


Figure 2. Scanning electron microscope images of (a,b) GaN-ReS₂, (c,d) ZnS-ReS₂ and (e,f) ZnO-ReS₂ *core-shell* nanowires at different magnifications.

Based on the TEM measurements, the optimal sulfurization temperature for obtaining high-quality monocrystalline ReS₂ layers from the magnetron-sputtered rhenium oxide coating was found to be 750-800°C. Fig. S3 shows as ReS₂ shell converted at 750°C that still exhibits a polycrystalline nature, as indicated by the non-parallel placement of the layers, while the TEM images in Fig. 1 shows a high degree of crystallinity at 800°C. In the case of ZnO NWs, a structural change in the NW could

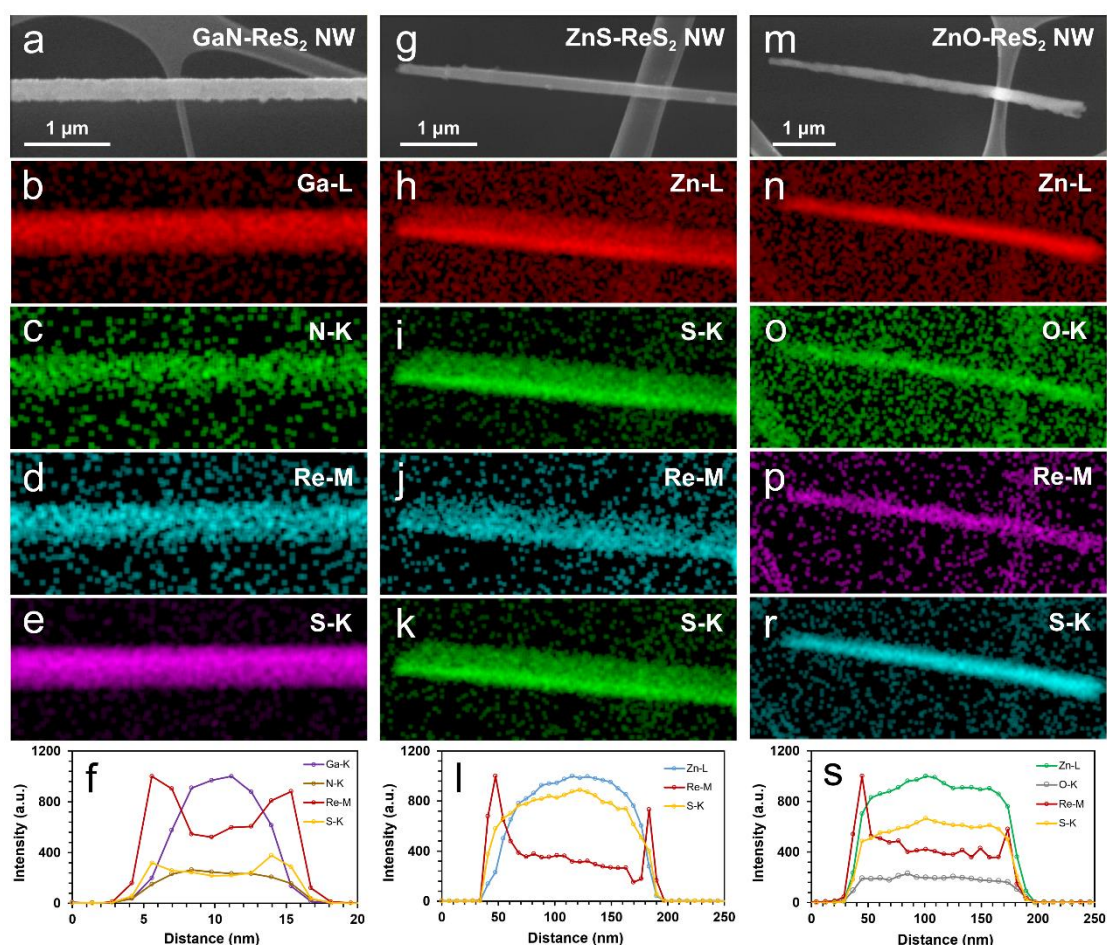


Figure 3. EDX elemental mapping of (a-f) GaN-ReS₂, (g-l) ZnS-ReS₂ and (m-s) ZnO-ReS₂ *core-shell* nanowires. The mapping was performed in SEM, and shows the uniform distribution of the ReS₂ shell along the nanowire, while cross-sectional line scans obtained via TEM confirm the distinct interfaces of the *core-shell* structure.

be observed around 750°C, and a further experiment was therefore performed. Fig. S4 shows TEM images of ZnO-ReS₂ NWs converted at 700°C and 750°C. It can clearly be seen that the ReS₂ is still polycrystalline at 700°C, and becomes monocrystalline at higher temperatures, while the ZnO core re-crystallizes at 750°C. It can be deduced that ZnO NWs undergo a sulfurization process at this temperature and are converted to the ZnS phase [22,60]. It is also worth noting that in our previous work [24], no detectable

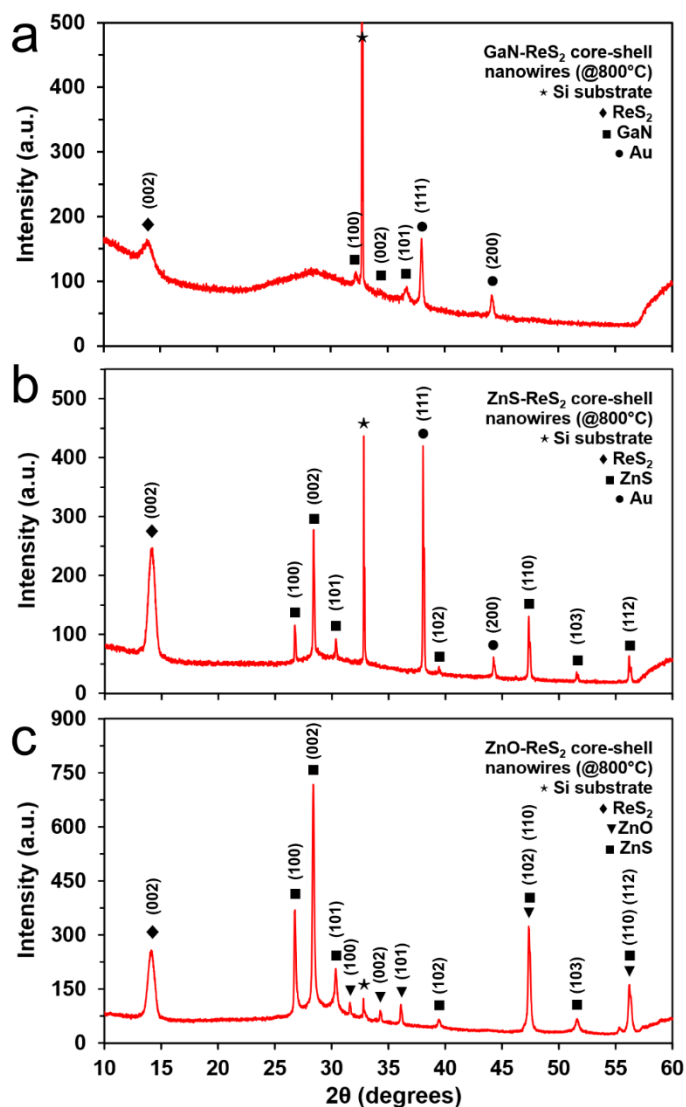


Figure 4. X-ray diffraction patterns of as-grown (a) GaN-ReS₂, (b) ZnS-ReS₂, and (c) ZnO-ReS₂ *core-shell* nanowire arrays on Si(100)/SiO₂ substrates, prepared at 800°C. Note the ZnS phase peaks in the ZnO-ReS₂ pattern, which emerge due to a partial or full conversion of ZnO to ZnS starting at around 750°C.

formation of the ZnS phase at 800°C was observed in ZnO-WS₂ *core-shell* NWs due to the unique growth mechanism of the WS₂ shell.

To confirm the presence of phases in the as-grown *core-shell* NW samples, XRD measurements were performed on the NW arrays (converted at 800°C) on Si(100)/SiO₂ substrates (see Fig. 4). All XRD patterns contained a strong ReS₂ (002) peak of the triclinic P-1 (No. 2) phase (ICDD-PDF #52-0818) as well as Bragg peaks, which were attributed to the Si(100) substrate (forbidden Si(200) reflection at $2\theta \approx 33^\circ$) and the gold nanoparticles used for VLS growth (ICDD-PDF #04-0784). For the GaN-ReS₂ and ZnS-ReS₂ NW samples, GaN (ICDD-PDF #50-0792) and ZnS (ICDD-PDF #36-1450) Bragg peaks are clearly apparent. For the ZnO-ReS₂ NWs samples, both ZnO (ICDD-PDF #36-1451) and ZnS phases were present, confirming that ZnO NWs are partly or fully converted to ZnS above 750°C, as observed in the TEM measurements. It should be noted that the ratio between the ReS₂ and NW XRD peak intensities is related not only to the amount of ReS₂ on the NWs, but also to that on the Si(100)/SiO₂ substrate, hence, it cannot be used to quantitatively describe the phase composition ratio of the *core-shell* heterostructures. Since the ZnO NWs undergo the sulfurization process before the monocrystalline ReS₂ shell is formed, this material may not be a suitable template for ReS₂ growth, although it was possible to grow high-quality monocrystalline layers of ReS₂ on GaN and ZnS NWs.

Additional XRD measurements were performed on as-prepared ReS₂ thin film samples in order to investigate the degree of crystallinity over a wider range of sulfurization temperatures. Fig. S5 shows XRD patterns for ReS₂ thin films, converted at 450°C, 700°C and 1000°C. The evolution of the ReS₂ (002) peak at 14.6° with increasing temperature can be clearly seen from the patterns, as the (004) and (006) peaks can be distinguished at very high temperatures [59]. This, therefore, confirms

that the conversion of a rhenium oxide film in sulfur vapour is a viable method for obtaining crystalline ReS₂ thin films. The Bragg peak at 33° is attributed to diffraction at the SiO₂/Si(100) substrate, and no other phases were observed. XRD measurements indicated that a small amount of crystalline ReS₂ phase starts to appear even at 450°C; however, the conversion rate becomes significant only at around 700°C.

Micro-Raman scattering spectra for as-grown ZnO-ReS₂, ZnS-ReS₂ and GaN-ReS₂ *core-shell* NWs (measured in each case from a single NW) were compared with those of bulk ReS₂ microcrystals in Fig. 5. There are 12 atoms in the unit cell of triclinic (P-1) ReS₂ that give rise to 36 Γ -point phonon modes, of which 18 symmetric A_g modes are Raman active and nondegenerate [61]: these are indicated by vertical lines in Fig. 5. All of the Raman active phonon modes were observed in bulk ReS₂ and *core-shell* NWs. The Raman bands are narrower for bulk ReS₂ and GaN-ReS₂ NWs, and are slightly broadened for ZnO-ReS₂ and ZnS-ReS₂ NWs. This broadening leads to an overlap of the nearest bands located around 280 cm⁻¹, 307 cm⁻¹ and 322 cm⁻¹. The most intense bands (E_g-like modes [61]) are located at about 152 cm⁻¹, 163 cm⁻¹ and 213 cm⁻¹, and mainly involve in-plane vibrations of Re atoms [61].

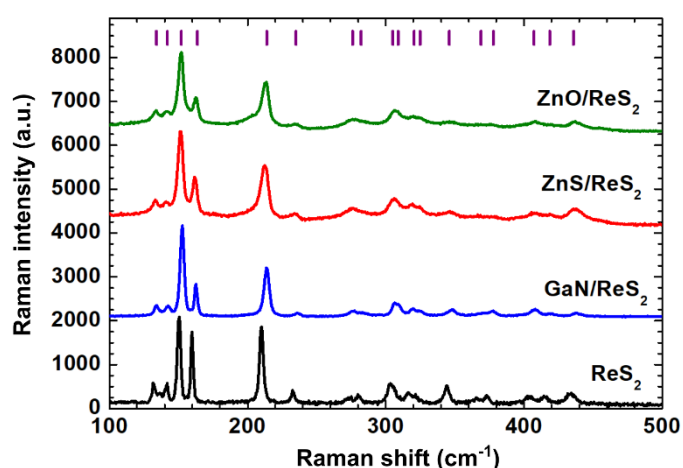


Figure 5. Micro-Raman scattering spectra for as-grown ZnO-ReS₂, ZnS-ReS₂ and GaN-ReS₂ *core-shell* nanowires, prepared at 800 °C, and bulk ReS₂. Vertical lines indicate the positions of the Raman active A_g phonon modes.

The XANES spectra for bulk ReS₂ and GaN-ReS₂, ZnS-ReS₂ and ZnO-ReS₂ *core-shell* NWs are close (Fig. 6). They are dominated by a strong “white line” at 10542 eV due to the 2p_{3/2}(Re)→5d(Re)-3p(S) dipole allowed transition. The low signal/noise ratio for the ZnO-ReS₂ sample is due to the small amount present. Note that the Ga K-edge XANES was also collected for GaN-ReS₂ NWs due to the close proximity of the Ga (10367 eV) and Re (10535 eV) absorption edges. This result is in agreement with that reported for GaN NWs in [62].

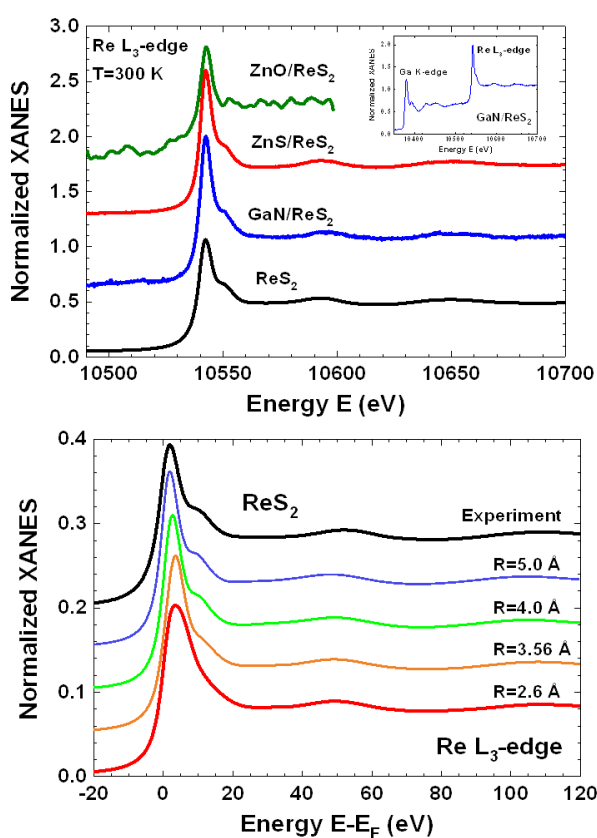


Figure 6. (a) X-ray absorption near edge structure (XANES) spectra for bulk ReS₂ and GaN-ReS₂, ZnS-ReS₂ and ZnO-ReS₂ *core-shell* nanowires measured at the Re L₃-edge. The inset shows the full acquired XANES spectrum for GaN-ReS₂ nanowires including the Ga K-edge, which is located about 160 eV below the Re L₃-edge. (b) Results of the *ab initio* Re L₃-edge XANES simulations for bulk ReS₂, performed for atomic clusters with radius R from 2.6 to 5.0 Å. The XANES spectra in (a) and (b) are vertically shifted for clarity.

The Re L₃-edge XANES for bulk ReS₂ agrees well with the data in the literature [63]. We interpreted it using the results of *ab initio* XANES calculations using the finite difference method (FDM) as implemented in the FDMNES code [64]. The XANES spectra were calculated for the ReS₂ crystallographic structure taken from [65] by averaging over four non-equivalent rhenium atoms in the unit cell. Relativistic calculations were performed using a self-consistent potential for atomic cluster radii up to 5 Å. The real energy-dependent exchange Hedin-Lundqvist potential was employed, and the calculated XANES spectra were broadened by natural core-level width ($\Gamma(\text{L}_3\text{-Re}) = 5.04$ eV). The results are shown in Fig. 6(a).

The results of the *ab initio* Re L₃-edge XANES simulations of bulk ReS₂ by the FDMNES code are shown in Fig. 6(b). It can be seen that a cluster with a radius of 4.0 Å around the absorbing rhenium atom is sufficient to reproduce the experimental XANES. The white line located at about 2 eV above the Fermi level E_F and the two maxima at about 50 and 110 eV in Fig. 6(b) are due to the interaction between the excited photoelectron and six sulfur atoms at 2.3-2.5 Å, which form the first coordination shell of the absorbing rhenium atom. The shoulder located at 10 eV appears after the addition of the second coordination shell, composed of four rhenium atoms at 2.8-3.6 Å, and it becomes more pronounced with the addition of subsequent outer shells. Thus, the Re L₃-edge XANES of ReS₂ is determined mainly by the local atomic structure, which explains its similarity in both *core-shell* NWs and in the bulk (Fig. 6(a)).

Fig. 7 compares the total and projected densities of states (PDOSs) calculated for [0001]-oriented monolayered (ML) and three-layered (3ML) ReS₂ nanosheets, as shown in Figs. 7(a) and 7(b), respectively, and the PDOSs calculated for slab models of [1-100] oriented ZnS, ZnO, and GaN NWs (Figs. 7(c), 7(d), and 7(e), respectively).

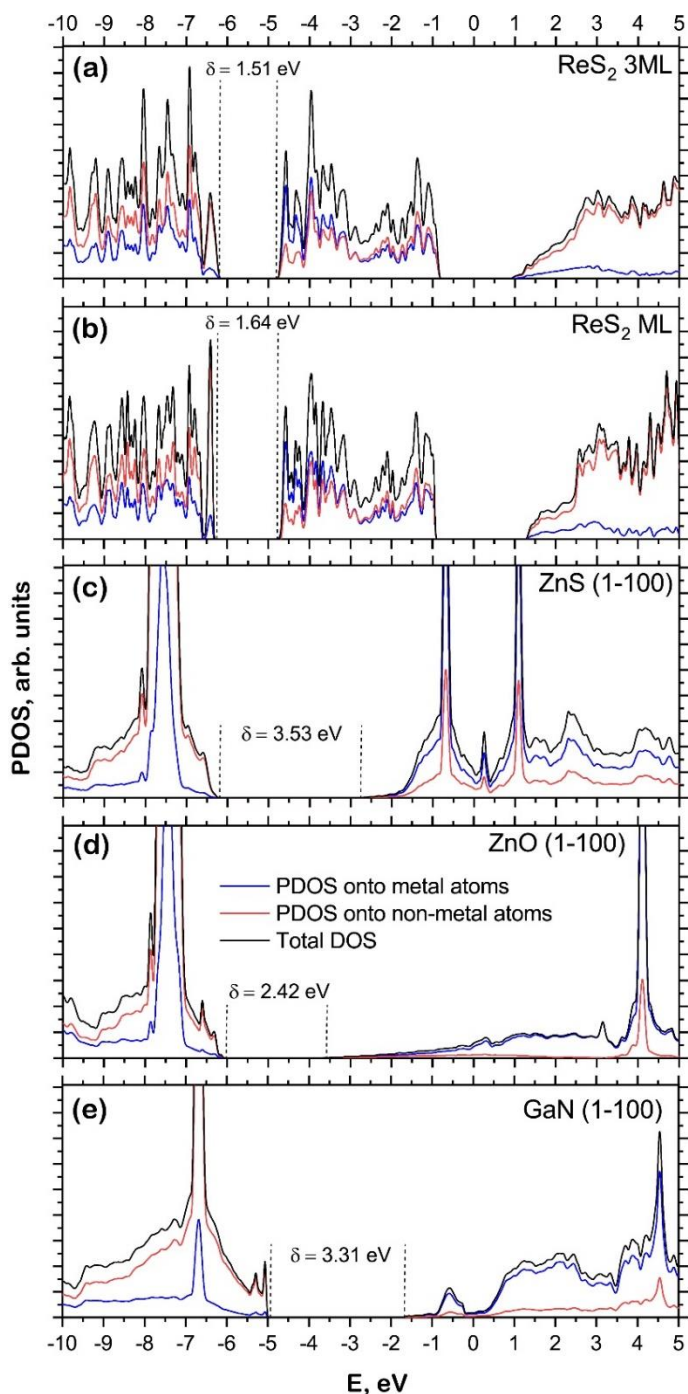


Figure 7. Total and projected densities of states (PDOSs) calculated for (a) 3-monolayer-thick ReS₂ nanosheet, (b) monolayer-thick ReS₂ nanosheet, (c) ZnS (1-100) slab, (d) ZnO (1-100) slab, and (e) GaN (1-100) slab. The 24-layer thick slabs are symmetrically terminated to mimic the surface of the nanowires. PDOSs are projected onto all orbitals of the corresponding metal or non-metal atoms. The energy scale is shown with respect to the vacuum level.

The PDOS calculated for bi-layered (2ML) ReS₂ nanosheet is not presented in Fig. 7, as its electronic structure does not differ in practise from that calculated for 3ML ReS₂.

For all of the ReS₂ nanosheets under study, the upper part of the valence band (VB) is predominantly formed by the S (3p) orbitals, with significant contributions from the Re (5d) orbitals. The bottom of the conduction band (CB) for the ReS₂ nanosheets is formed mainly of the Re (5d) states. In the case of the ZnS, ZnO and GaN (1-100) surfaces, the top of the VB is formed of the O or N 2p states, while the 3d orbitals of Zn and Ga mainly form the bottom of the CB. It can clearly be seen that the band edge positions of ML and 3ML ReS₂ differ from those of the bulk (Fig. S6(a)). The bottom of the CB is shifted down, closer to the hydrogen evolution potential of -4.44 eV, while the position of the top of the VB calculated for the (1-100) surfaces of ZnO, ZnS, and GaN NWs is located near the oxygen reduction potential of -5.67 eV. The band alignment is shown schematically in Fig. 8. This allows us to conclude that the hybrid

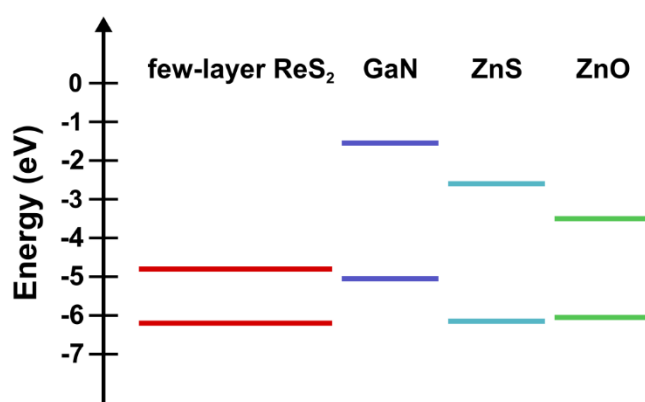


Figure 8. Schematic representation of the heterostructure band alignment, based on the DFT calculations performed here.

core-shell ZnO-ReS₂, ZnS-ReS₂, and GaN-ReS₂ NWs can be considered promising potential materials for efficient solar-driven photo(electro)catalytic water splitting.

4. Conclusions

In this study, we have demonstrated the possibility of using two-step growth process for few-layer ReS₂ on different semiconductor material NWs. An amorphous

rhodium oxide coating was deposited on the substrates by reactive DC magnetron sputtering, followed by sulfurization of the pre-deposited film at different temperatures. High-quality monocrystalline few-layer ReS₂ was successfully grown on GaN and ZnS NWs at a sulfurization temperature of 800°C, although it was only possible to obtain a polycrystalline ReS₂ shell on ZnO NWs at lower temperatures, due to the conversion of ZnO to the ZnS phase at 750°C. The obtained *core-shell* heterostructures were characterized in detail using XRD, SEM, TEM, EDX, micro-Raman spectroscopy and X-ray absorption spectroscopy, and experimental characterizations were supported by total energy calculations of the electronic structure for the studied materials. We showed that in a similar way to other TMDs such as MoS₂ and WS₂, few-layer ReS₂ can be obtained by sulfurization of a pre-deposited rhodium oxide film. Our results demonstrate the potential of using NWs as a template for the growth of layered vdW materials to produce novel hybrid *core-shell* nanostructures that have a number of potential applications, such as in electronics, optoelectronics and various energy applications.

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Supplementary information

Supplementary information is available containing XRD and SEM data for the pure NWs used in the experiments and the ReS₂ thin films, SAED patterns and analysis of the as-grown *core-shell* NWs, TEM images of GaN-ReS₂ and ZnO-ReS₂ NWs at different annealing temperatures, and PDOSs calculated for the bulk phases of ReS₂, ZnS, ZnO, and GaN crystals.

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